

# Synthesis of 4*H*,5*H*-pyrano[3,2-*c*]pyran-5-ones in aqueous media

Da-Qing Shi<sup>a,b,\*</sup>, Li-Hui Niu<sup>a</sup>, Xiang-Shan Wang<sup>a,b</sup>, Qi-Ya Zhuang<sup>a,b</sup> and Yong Zhang<sup>c</sup>

<sup>a</sup>Department of Chemistry, Xuzhou Normal University, Xuzhou 221116, P. R. China

<sup>b</sup>The Key Laboratory of Biotechnology on Medical Plants of Jiangsu Province, Xuzhou 221116, P. R. China

<sup>c</sup>School of Chemistry and Chemical Engineering, Suzhou University, Suzhou 215006, P. R. China

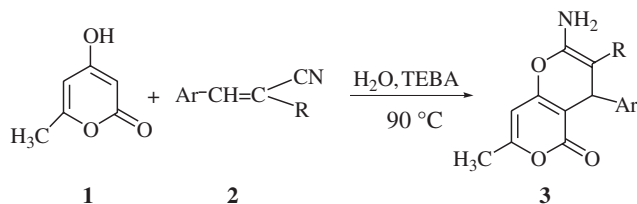
The reaction of 4-hydroxy-6-methylpyran-2-one with substituted cinnamitriles in water in the presence of triethylbenzylammonium chloride (TEBA) provide an efficient route to 2-amino-4-aryl-4*H*,5*H*-pyrano[3,2-*c*]pyran-5-one derivatives. The products were characterised by IR, <sup>1</sup>H NMR, elemental analysis and were further confirmed by the X-ray crystal structure analysis.

**Keywords:** 4*H*,5*H*-pyrano[3,2-*c*] pyran-5-one, reaction in water, green chemistry

The need to reduce the amount of toxic waste and by products arising from chemical processes requires increasing emphasis on the use of less toxic and environmentally compatible materials in the design of new synthetic methods.<sup>1–3</sup> One of the most promising approaches uses water as a reaction medium.<sup>4–6</sup> Breslow,<sup>7–8</sup> who showed that hydrophobic effects could strongly enhance the rate of several organic reactions, rediscovered the use of water as a solvent in organic reactions in the 1980s. In recent years, there has been increasing recognition that water is an attractive medium for many organic reactions.<sup>9–14</sup> The aqueous medium is cheaper, safer and environmentally friendly in comparison with organic solvents. Generally, the low solubility<sup>15</sup> of most reagents in water is not an obstacle to the reactivity, which, on the contrary, is reduced by the use of cosolvents.

In recent years, the synthesis of benzopyran derivatives has attracted great interest. Since the discovery of cromakalim as a typical ATP-sensitive potassium channel opener (PCO), a large number of benzopyran derivatives have been synthesised and demonstrated to possess potent relaxant activity on blood vessels, cardiac muscles, and other smooth muscles.<sup>16–20</sup> These agents may find use in the treatment of a variety of diseases such as hypertension, asthma, ischemia and urinary incontinence. Benzopyrans with amino and cyano groups are also synthons of some special natural products.<sup>21–22</sup> Benzopyran derivatives are generally prepared by the reaction of substituted cinnamitriles and active methylene carbonyl compounds in an organic solvent (*e.g.* ethanol) in the presence of organic bases like piperidine or triethylamine.<sup>23–25</sup> Based on our previous studies on the use of water as solvent for carrying out carbon–carbon bond forming reactions under heterogeneous catalysis,<sup>26–34</sup> we now report the synthesis of 4*H*,5*H*-pyrano[3,2-*c*] pyran-5-ones in aqueous media.

When 4-hydroxy-6-methylpyran-2-one (**1**) and substituted cinnamitriles (**2**) were stirred for 4–15 h at 90 °C in aqueous suspension in the presence of triethylbenzyl-ammonium chloride (TEBA), 2-amino-4-aryl-4*H*,5*H*-pyrano[3,2-*c*]pyran-5-ones (**3**) were obtained in excellent yields (see Scheme). The results are showed in Table 1.



**Scheme**

**Table 1** Synthetic results for compounds **3**

Entry	Ar	R	Time/h	Yield/% <sup>a</sup>
<b>3a</b>	4-ClC <sub>6</sub> H <sub>4</sub>	CN	7	80
<b>3b</b>	4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	CN	8	91
<b>3c</b>	2-ClC <sub>6</sub> H <sub>4</sub>	CN	6	98
<b>3d</b>	2,4-Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CN	5	91
<b>3e</b>	4-BrC <sub>6</sub> H <sub>4</sub>	CN	8	95
<b>3f</b>	3,4-OCH <sub>2</sub> OC <sub>6</sub> H <sub>3</sub>	CN	12	95
<b>3g</b>	4-FC <sub>6</sub> H <sub>4</sub>	CN	4	88
<b>3h</b>	3,4-(CH <sub>3</sub> O) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CN	10	91
<b>3i</b>	3-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	CN	6	86
<b>3j</b>	3,4-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CN	8	75
<b>3k</b>	2,4-Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CO <sub>2</sub> Et	12	77
<b>3l</b>	4-CH <sub>2</sub> OC <sub>6</sub> H <sub>4</sub>	CO <sub>2</sub> Et	10	87
<b>3m</b>	4-FC <sub>6</sub> H <sub>4</sub>	CO <sub>2</sub> Et	15	83
<b>3n</b>	2-ClC <sub>6</sub> H <sub>4</sub>	CO <sub>2</sub> Et	7	76
<b>3o</b>	C <sub>6</sub> H <sub>5</sub>	CO <sub>2</sub> Et	10	95
<b>3p</b>	4-ClC <sub>6</sub> H <sub>4</sub>	CO <sub>2</sub> Et	15	76
<b>3q</b>	C <sub>6</sub> H <sub>5</sub>	CO <sub>2</sub> Me	12	75
<b>3r</b>	4-ClC <sub>6</sub> H <sub>4</sub>	CO <sub>2</sub> Me	12	84

<sup>a</sup>Isolated yield based on the amount of 4-hydroxy-6-methylpyran-2-one **1**.

Table 1 shows the results using a series of aromatic aldehydes that undergo the reaction to give excellent yields (75–98 %) of the products. This procedure does not require the use of any organic solvent. In fact the target compounds **3** were isolated in a practically pure form by simple Büchner filtration of the final aqueous mixture. In this reaction TEBA is necessary. If no TEBA is added the reaction takes a long time and the yield will very low. The role of the TEBA is to form micelles.

Product structures were established on the basis of spectroscopic data, particularly <sup>1</sup>H NMR analysis, and were further confirmed by the X-ray crystal structure analysis of the products **3f**<sup>35</sup> (Fig. 1) and **3l**<sup>36</sup> (Fig. 2).

In summary, the conversion of substituted cinnamitriles and 4-hydroxy-6-methylpyran-2-one into 2-amino-4-aryl-4*H*,5*H*-pyrano[3,2-*c*]pyran-5-ones can be efficiently performed in water as a solvent using a catalytic amount of TEBA. Compared to the previous methods in an organic solvent,<sup>37</sup> this new protocol has the advantages of good yields, low cost, and simple operation.

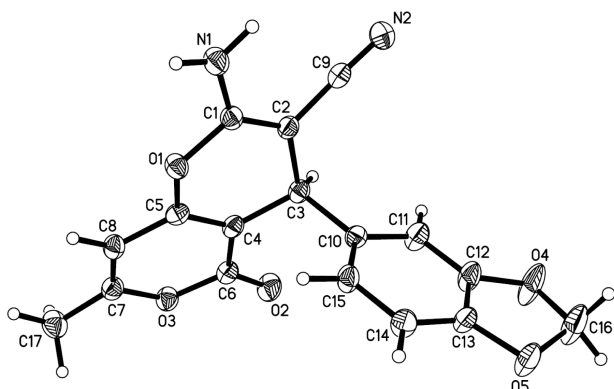
## Experimental

IR spectra were recorded on a FT IR-8101 spectrometer in KBr. <sup>1</sup>H NMR spectra were determined on a Inova-400 MHz spectrometer using DMSO-*d*<sub>6</sub> solutions. Chemical shifts are expressed in ppm downfield from internal tetramethylsilane. Microanalyses were carried out on a Perkin-Elmer 2400 II elemental analyser. X-ray diffraction was recorded using a Rigaku Mercury diffractometer.

*General procedure for the synthesis of 2-amino-4-aryl-4H,5H-pyrano[3,2-*c*]pyran-5-one derivatives (3)*

A mixture of 4-hydroxy-6-methylpyran-2-one **1** (2 mmol), the substituted cinnamitrile **2** (2 mmol) and TEBA (0.15 g, 0.66 mmol) in H<sub>2</sub>O (10 ml) was stirred for 4–15 h at 90 °C and then cooled to

\* Correspondent. E-mail: dqshi@263.net



**Fig. 1** The X-ray crystal structure of compound **3f**.

room temperature. The solid material formed was collected by filtration, washed with water and recrystallised from 95 % ethanol to give pure **3**.

For **3a**, **3b**, **3l**, **3p** and **3r** the AA'XX' systems of the aromatic groups give apparent pairs of doublets in the  $^1\text{H}$  NMR spectra.

**2-Amino-3-cyano-4-(4-chlorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3a)**:<sup>37</sup> M.p. 230–231 °C; IR:  $\nu/\text{cm}^{-1}$  3384, 3324, 3031, 2201, 1711, 1673, 1645, 1614, 1585, 1489, 1445, 1413, 1384, 1261, 1195, 1142, 1094, 1042, 1009, 981, 829, 778;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.23 (3H, s, CH<sub>3</sub>), 4.32 (1H, s, C<sup>4</sup>-H), 6.28 (1H, s, C<sup>8</sup>-H), 7.22 (2H, d,  $J$  = 8.8 Hz, ArH), 7.24 (2H, s, NH<sub>2</sub>), 7.37 (2H, d,  $J$  = 8.8 Hz, ArH); Anal. calcd for C<sub>16</sub>H<sub>11</sub>ClN<sub>2</sub>O<sub>3</sub>: C 61.1, H 3.5, N 8.9; found C 61.2, H 3.4, N 8.9.

**2-Amino-3-cyano-4-(4-nitrophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3b)**: M.p. 219–220 °C; IR:  $\nu/\text{cm}^{-1}$  3398, 3297, 3083, 2203, 1707, 1670, 1642, 1605, 1518, 1445, 1412, 1379, 1332, 1264, 1198, 1178, 1138, 1111, 1044, 984, 872, 823, 742, 693;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.24 (3H, s, CH<sub>3</sub>), 4.51 (1H, s, C<sup>4</sup>-H), 6.32 (1H, s, C<sup>8</sup>-H), 7.35 (2H, s, NH<sub>2</sub>), 7.51 (2H, d,  $J$  = 6.8 Hz, ArH), 8.18 (2H, d,  $J$  = 6.8 Hz, ArH); Anal. calcd for C<sub>16</sub>H<sub>11</sub>N<sub>3</sub>O<sub>5</sub>: C 59.1, H 3.4, N 12.9; found C 59.2, H 3.25, N 12.8.

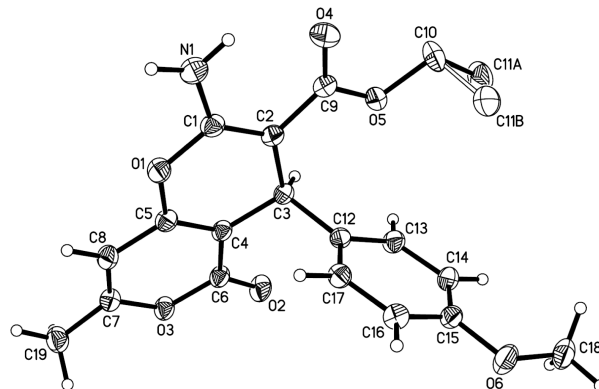
**2-Amino-3-cyano-4-(2-chlorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3c)**: M.p. 267–268 °C; IR:  $\nu/\text{cm}^{-1}$  3471, 3344, 2141, 1700, 1669, 1638, 1602, 1579, 1473, 1443, 1376, 1322, 1262, 1196, 1180, 1139, 1043, 982, 963, 820, 762;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.24 (3H, s, CH<sub>3</sub>), 4.80 (1H, s, C<sup>4</sup>-H), 6.29 (1H, s, C<sup>8</sup>-H), 7.22 (2H, s, NH<sub>2</sub>), 7.23–7.32 (3H, m, ArH), 7.38–7.41 (1H, m, ArH); Anal. calcd for C<sub>16</sub>H<sub>11</sub>ClN<sub>2</sub>O<sub>3</sub>: C 61.1, H 3.5, N 8.9; found C 61.2, H 3.4, N 8.7.

**2-Amino-3-cyano-4-(2,4-dichlorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3d)**: M.p. 230–231 °C; IR:  $\nu/\text{cm}^{-1}$  3347, 3141, 2972, 2921, 2197, 1715, 1669, 1640, 1611, 1472, 1388, 1265, 1200, 1179, 1103, 1042, 975, 859, 813, 790, 737;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.24 (3H, s, CH<sub>3</sub>), 4.80 (1H, s, C<sup>4</sup>-H), 6.29 (1H, s, C<sup>8</sup>-H), 7.27 (2H, s, NH<sub>2</sub>), 7.29 (1H, d,  $J$  = 8.8 Hz, ArH), 7.38 (1H, m,  $J$  = 8.8 Hz, ArH), 7.55 (1H, s, ArH); Anal. calcd for C<sub>16</sub>H<sub>10</sub>Cl<sub>2</sub>N<sub>2</sub>O<sub>3</sub>: C 55.0, H 2.9, N 8.0; found C 55.3, H 2.6, N 8.1.

**2-Amino-3-cyano-4-(4-bromophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3e)**: M.p. 223–224 °C; IR:  $\nu/\text{cm}^{-1}$  3388, 3325, 2201, 1708, 1680, 1644, 1609, 1486, 1445, 1408, 1385, 1314, 1283, 1261, 1196, 1176, 1142, 1098, 1074, 1043, 982, 854, 829, 776;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.30 (3H, s, CH<sub>3</sub>), 4.31 (1H, s, C<sup>4</sup>-H), 6.28 (1H, s, C<sup>8</sup>-H), 7.17 (2H, d,  $J$  = 8.4 Hz, ArH), 7.25 (2H, s, NH<sub>2</sub>), 7.51 (2H, d,  $J$  = 8.4 Hz, ArH); Anal. calcd for C<sub>16</sub>H<sub>11</sub>BrN<sub>2</sub>O<sub>3</sub>: C 53.5, H 3.1, N 7.8; found C 53.7, H 2.9, N 8.0.

**2-Amino-3-cyano-4-(3,4-methylenedioxyphenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3f)**: M.p. 228–230 °C; IR:  $\nu/\text{cm}^{-1}$  3400, 3328, 2885, 2202, 1706, 1670, 1644, 1613, 1592, 1502, 1485, 1443, 1382, 1259, 1193, 1139, 1038, 984, 922, 796, 784;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.23 (3H, s, CH<sub>3</sub>), 4.22 (1H, s, C<sup>4</sup>-H), 5.99 (2H, s, OCH<sub>2</sub>O), 6.26 (1H, s, C<sup>8</sup>-H), 6.66 (1H, d,  $J$  = 8.0 Hz, ArH), 6.72 (1H, s, ArH), 6.83 (1H, m,  $J$  = 8.0 Hz, ArH), 7.16 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>17</sub>H<sub>12</sub>N<sub>2</sub>O<sub>5</sub>: C 63.0, H 3.7, N 8.6; found C 62.7, H 3.5, N 8.85.

**2-Amino-3-cyano-4-(4-fluorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3g)**:<sup>37</sup> M.p. 242–243 °C; IR:  $\nu/\text{cm}^{-1}$  3399, 3326, 2885, 2197, 1711, 1672, 1644, 1613, 1508, 1445, 1415, 1383, 1323, 1261, 1220, 1177, 1139, 1097, 1043, 980, 854, 827, 785, 747;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.22 (3H, s, CH<sub>3</sub>), 4.32 (1H, s, C<sup>4</sup>-H), 6.27



**Fig. 2** The X-ray crystal structure of compound **3l**. The ethyl group is disordered.

(1H, s, C<sup>8</sup>-H), 7.11–7.15 (2H, m, ArH), 7.21 (2H, s, NH<sub>2</sub>), 7.22–7.26 (2H, m, ArH); Anal. calcd for C<sub>16</sub>H<sub>11</sub>N<sub>2</sub>O<sub>3</sub>: C 64.4, H 3.7, N 9.4; found C 64.7, H 3.5, N 9.5.

**2-Amino-3-cyano-4-(3,4-dimethoxyphenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3h)**: M.p. 207–209 °C; IR:  $\nu/\text{cm}^{-1}$  3397, 3328, 2962, 2197, 1708, 1666, 1643, 1611, 1515, 1463, 1422, 1380, 1262, 1143, 1021, 969, 854, 819, 761;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.23 (3H, s, CH<sub>3</sub>), 3.73 (6H, s, 2×CH<sub>3</sub>O), 4.24 (1H, s, C<sup>4</sup>-H), 6.26 (1H, s, C<sup>8</sup>-H), 6.67 (1H, d,  $J$  = 8.4 Hz, ArH), 6.78 (1H, s, ArH), 6.89 (1H, d,  $J$  = 8.4 Hz, ArH), 7.14 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>18</sub>H<sub>16</sub>N<sub>2</sub>O<sub>5</sub>: C 63.5, H 4.7, N 8.2; found C 63.8, H 4.5, N 8.5.

**2-Amino-3-cyano-4-(3-nitrophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3i)**: M.p. 234–235 °C; IR:  $\nu/\text{cm}^{-1}$  3400, 3327, 2199, 1717, 1673, 1646, 1610, 1526, 1477, 1448, 1378, 1344, 1263, 1200, 1143, 1095, 1042, 977, 822, 733, 685;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.24 (3H, s, CH<sub>3</sub>), 4.57 (1H, s, C<sup>4</sup>-H), 6.32 (1H, s, C<sup>8</sup>-H), 7.35 (2H, s, NH<sub>2</sub>), 7.62–7.66 (1H, m, ArH), 7.72–7.75 (1H, m, ArH), 8.05–8.06 (1H, m, ArH), 8.11–8.14 (1H, m, ArH); Anal. calcd for C<sub>16</sub>H<sub>11</sub>N<sub>3</sub>O<sub>5</sub>: C 59.1, H 3.4, N 12.9; found C 59.3, H 3.2, N 12.9.

**2-Amino-3-cyano-4-(3,4-dimethylphenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-5-one (3j)**: M.p. 228–230 °C; IR:  $\nu/\text{cm}^{-1}$  3417, 3324, 3098, 2920, 2200, 1706, 1678, 1644, 1644, 1613, 1587, 1503, 1446, 1413, 1379, 1260, 1195, 1135, 1091, 1038, 964, 819, 791, 780;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  2.17 (3H, s, CH<sub>3</sub>), 2.19 (3H, s, CH<sub>3</sub>), 2.22 (3H, s, CH<sub>3</sub>), 4.18 (1H, s, C<sup>4</sup>-H), 6.26 (1H, s, C<sup>8</sup>-H), 6.68 (2H, d,  $J$  = 8.4 Hz, ArH), 6.92 (1H, s, ArH), 7.05 (1H, d,  $J$  = 8.4 Hz, ArH), 7.13 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>18</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>: C 70.1, H 5.2, N 9.1; found C 70.3, H 5.0, N 9.3.

**Ethyl 2-amino-5-oxo-4-(2,4-dichlorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3k)**: M.p. 204–205 °C; IR:  $\nu/\text{cm}^{-1}$  3435, 3299, 3064, 2985, 1714, 1679, 1614, 1504, 1471, 1380, 1289, 1174, 1079, 1046, 956, 863, 849, 731;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  1.04 (3H, t,  $J$  = 7.2 Hz, CH<sub>3</sub>), 2.22 (3H, s, CH<sub>3</sub>), 3.93 (2H, q,  $J$  = 7.2 Hz, CH<sub>2</sub>O), 4.88 (1H, s, C<sup>4</sup>-H), 6.18 (1H, s, C<sup>8</sup>-H), 7.25 (1H, d,  $J$  = 8.4 Hz, ArH), 7.31 (1H, d,  $J$  = 8.4 Hz, ArH), 7.42 (1H, m, ArH), 7.83 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>18</sub>H<sub>15</sub>Cl<sub>2</sub>NO<sub>5</sub>: C 54.6, H 3.8, N 3.5; found C 54.8, H 3.7, N 3.7.

**Ethyl 2-amino-5-oxo-4-(4-methoxyphenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3l)**:<sup>37</sup> M.p. 173–174 °C; IR:  $\nu/\text{cm}^{-1}$  3426, 3296, 3064, 3088, 2984, 1715, 1684, 1616, 1508, 1457, 1379, 1290, 1172, 1082, 1027, 957, 827, 735;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  1.09 (3H, t,  $J$  = 7.2 Hz, CH<sub>3</sub>), 2.21 (3H, s, CH<sub>3</sub>), 3.69 (3H, s, CH<sub>3</sub>O), 3.97 (2H, q,  $J$  = 7.2 Hz, CH<sub>2</sub>O), 4.48 (1H, s, C<sup>4</sup>-H), 6.28 (1H, s, C<sup>8</sup>-H), 6.79 (2H, d,  $J$  = 8.8 Hz, ArH), 7.07 (2H, d,  $J$  = 8.8 Hz, ArH), 7.66 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>19</sub>H<sub>19</sub>NO<sub>6</sub>: C 63.9, H 5.4, N 3.9; found C 63.95, H 5.1, N 3.8.

**Ethyl 2-amino-5-oxo-4-(4-fluorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3m)**: M.p. 162–163 °C; IR:  $\nu/\text{cm}^{-1}$  3434, 3307, 3099, 3088, 2984, 1708, 1681, 1625, 1504, 1382, 1290, 1250, 1221, 1141, 1095, 834, 743;  $^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  1.06 (3H, t,  $J$  = 7.2 Hz, CH<sub>3</sub>), 2.22 (3H, s, CH<sub>3</sub>), 3.96 (2H, q,  $J$  = 7.2 Hz, CH<sub>2</sub>O), 4.53 (1H, s, C<sup>4</sup>-H), 6.29 (1H, s, C<sup>8</sup>-H), 7.03–7.08 (2H, m, ArH), 7.18–7.21 (2H, m, ArH), 7.73 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>18</sub>H<sub>16</sub>FN<sub>2</sub>O<sub>5</sub>: C 62.6, H 4.7, N 4.1; found C 62.85, H 4.5, N 4.2.

**Ethyl 2-amino-5-oxo-4-(2-chlorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3n)**: M.p. 203–205 °C; IR:  $\nu/\text{cm}^{-1}$  3422, 3296, 3093, 2988, 1709, 1687, 1617, 1511, 1440, 1378, 1286, 1250, 1176, 1143, 1077, 1039, 956, 815, 773;  $^1\text{H}$  NMR (DMSO- $d_6$ )

$\delta$ : 1.03 (3H, t,  $J = 7.2$  Hz, CH<sub>3</sub>), 2.21 (3H, s, CH<sub>3</sub>), 3.91 (2H, q,  $J = 7.2$  Hz, CH<sub>2</sub>O), 4.90 (1H, s, C<sup>4</sup>-H), 6.27 (1H, m, ArH), 7.18–7.29 (4H, m, ArH), 7.78 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>18</sub>H<sub>16</sub>ClNO<sub>5</sub>: C 59.8, H 4.5 N 3.9; found C 60.0, H 4.25, N 3.7.

*Ethyl 2-amino-5-oxo-4-phenyl-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3o)*:<sup>37</sup> M.p. 182–183 °C; IR:  $\nu/\text{cm}^{-1}$  3428, 3304, 3094, 2986, 1710, 1687, 1612, 1507, 1443, 1382, 1286, 1213, 1140, 1079, 957, 817, 753, 703; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>)  $\delta$ : 1.07 (3H, t,  $J = 7.2$  Hz, CH<sub>3</sub>), 2.21 (3H, s, CH<sub>3</sub>), 3.96 (2H, q,  $J = 7.2$  Hz, CH<sub>2</sub>O), 4.54 (1H, s, C<sup>4</sup>-H), 6.29 (1H, s, C<sup>8</sup>-H), 7.12–7.26 (5H, m, ArH), 7.71 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>18</sub>H<sub>17</sub>NO<sub>5</sub>: C 66.05, H 5.2, N 4.3; found C 66.2, H 5.0, N 4.4.

*Ethyl 2-amino-5-oxo-4-(4-chlorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3p)*:<sup>38</sup> M.p. 156–157 °C; IR:  $\nu/\text{cm}^{-1}$  3431, 3307, 3064, 2980, 1718, 1688, 1618, 1504, 1442, 1410, 1375, 1290, 1213, 1140, 1076, 1013, 980, 955, 818, 758, 703; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>)  $\delta$ : 1.07 (3H, t,  $J = 7.2$  Hz, CH<sub>3</sub>), 2.22 (3H, s, CH<sub>3</sub>), 3.96 (2H, q,  $J = 7.2$  Hz, CH<sub>2</sub>O), 4.52 (1H, s, C<sup>4</sup>-H), 6.30 (1H, s, C<sup>8</sup>-H), 7.19 (2H, d,  $J = 8.4$  Hz, ArH), 7.30 (2H, d,  $J = 8.4$  Hz, ArH), 7.75 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>18</sub>H<sub>16</sub>ClNO<sub>5</sub>: C 59.8, H 4.5, N 3.9; found C 59.9, H 4.3, N 3.7.

*Methyl 2-amino-5-oxo-4-phenyl-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3q)*: M.p. 195–196 °C; IR:  $\nu/\text{cm}^{-1}$  3436, 3307, 3093, 3001, 1709, 1691, 1614, 1502, 1436, 1378, 1382, 1293, 1214, 1194, 1140, 1079, 980, 816, 729, 702; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>)  $\delta$ : 2.21 (3H, s, CH<sub>3</sub>), 3.52 (3H, s, CH<sub>3</sub>O), 4.54 (1H, s, C<sup>4</sup>-H), 6.29 (1H, s, C<sup>8</sup>-H), 7.16–7.24 (5H, m, ArH), 7.72 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>17</sub>H<sub>15</sub>NO<sub>5</sub>: C 65.2, H 4.8, N 4.5; found C 65.4, H 4.7, N 4.6.

*Methyl 2-amino-5-oxo-4-(4-chlorophenyl)-7-methyl-4H,5H-pyrano[3,2-c]pyran-3-carboxylate (3r)*: M.p. 145–146 °C; IR:  $\nu/\text{cm}^{-1}$  3401, 3295, 3087, 2946, 1734, 1690, 1618, 1523, 1489, 1435, 1378, 1381, 1267, 1196, 1137, 1086, 1013, 970, 936, 827, 783; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>)  $\delta$ : 2.22 (3H, s, CH<sub>3</sub>), 3.51 (3H, s, CH<sub>3</sub>O), 4.53 (1H, s, C<sup>4</sup>-H), 6.29 (1H, s, C<sup>8</sup>-H), 7.19 (2H, d,  $J = 8.4$  Hz, ArH), 7.29 (2H, d,  $J = 8.4$  Hz, ArH), 7.76 (2H, s, NH<sub>2</sub>); Anal. calcd for C<sub>17</sub>H<sub>14</sub>ClNO<sub>5</sub>: C 58.7, H 4.1, N 4.0; found C 59.0, H 4.0, N 3.95.

We are grateful to the “Surpassing Project” Foundation of Jiangsu Province and the Foundation of the Key Laboratory of Biotechnology on Medical Plants of Jiangsu Province for financial support.

Received 12 February 2005; accepted 9 June 2005  
Paper 05/3090

## References

- 1 J. Amato, *Science*, 1993, **259**, 1538.
- 2 D.L. Illman, *Chem. Eng. News*, 1993, **71**, 5.
- 3 D.L. Illman, *Chem. Eng. News*, 1994, **72**, 22.
- 4 C.J. Li and T.H. Chang, *Organic Reactions in Aqueous Media*, Wiley, New York, 1997.
- 5 F. Fringuelli, O. Piematti, F. Pizzo and L. Vaccaro, *Eur. J. Org. Chem.*, 2001, 439.
- 6 S.C. Stinson, *Chem. Eng. News*, 1996, **74**, 39.
- 7 R. Breslow and D.C. Rideout, *J. Am. Chem. Soc.*, 1980, **102**, 7816.
- 8 R. Breslow, *Acc. Chem. Rev.*, 1993, **93**, 2023.
- 9 C.J. Li, *Chem. Rev.*, 1993, **93**, 2023.
- 10 R. Ballini and G. Bosica, *Tetrahedron Lett.*, 1996, **37**, 8027.
- 11 R. Ballini, G. Bosica and T. Mecozzi, *Tetrahedron*, 1997, **53**, 7341.
- 12 A. Meijer, S. Otto and J.B.F.N. Engberts, *J. Org. Chem.*, 1998, **63**, 8589.
- 13 F. Bigi, L. Chesini, R. Maggi and G. Sartori, *J. Org. Chem.*, 1999, **64**, 1033.
- 14 F. Bigi, S. Carloni, L. Ferrari, R. Maggi, A. Mazzacani and G. Sartori, *Tetrahedron Lett.*, 2001, **42**, 5203.
- 15 F. Fringuelli, G. Pani, O. Piematti and F. Pizzo, *Tetrahedron*, 1994, **50**, 11499.
- 16 R.H. Poyser and T.C. Hamilton, *Drugs Future*, 1994, **19**, 39.
- 17 J.R. Empfield and K. Russell, *Ann. Rep. Med. Chem.*, 1996, **30**, 81.
- 18 B. Piroette, J. Fontaine and P. Lebrun, *Curr. Med. Chem.*, 1995, **2**, 573.
- 19 K.S. Atwal, *Curr. Med. Chem.*, 1996, **3**, 27.
- 20 J.M. Evans, T.C. Hamilton, S.D. Longman and G. Stemp, *Potassium Channels and their Modulators: From Synthesis to Clinical Experience*, Taylor & Francis, 1996.
- 21 C.N. O'Callaghan and T.B.H. McMarry, *J. Chem. Res., (S)*, 1995, 214.
- 22 S. Hatakeyama, N. Ochi, H. Numata and S. Takano, *J. Chem. Soc., Chem. Commun.*, 1998, 1202.
- 23 A.M. El-Agrody, M.S. El-Latif, A.H. Fakery and A.H. Bedair, *J. Chem. Res., (S)*, 2000, 26.
- 24 X.S. Wang, D.Q. Shi, S.J. Tu, C.S. Yao and Y.C. Wang, *Chin. J. Struct. Chem.*, 2002, **21**, 146.
- 25 A.Z. Sayed, N.A. El-Hady and A-E. El-Agrody, *J. Chem. Res., (S)*, 2000, 164.
- 26 D.Q. Shi, J. Chen, Q.Y. Zhuang, X.S. Wang and H.W. Hu, *J. Chem. Res., (S)* 2003, 674.
- 27 D.Q. Shi, J. Chen, Q.Y. Zhuang, X.S. Wang and H.W. Hu, *Chin. Chem. Lett.*, 2003, **14**, 1242.
- 28 D.Q. Shi, S. Zhang, Q.Y. Zhuang, S.J. Tu and H.W. Hu, *Chin. J. Org. Chem.*, 2003, **21**, 680.
- 29 D.Q. Shi, S. Zhang, Q.Y. Zhuang, S.J. Tu and H.W. Hu, *Chin. J. Org. Chem.*, 2003, **23**, 809.
- 30 D.Q. Shi, S. Zhang, Q.Y. Zhuang, S.J. Tu and H.W. Hu, *Chin. J. Org. Chem.*, 2003, **23**, 877.
- 31 D.Q. Shi, S. Zhang, Q.Y. Zhuang, S.J. Tu and H.W. Hu, *Chin. J. Org. Chem.*, 2003, **23**, 1036.
- 32 D.Q. Shi, S. Zhang, Q.Y. Zhuang, S.J. Tu and H.W. Hu, *Chin. J. Org. Chem.*, 2003, **23**, 1314.
- 33 D.Q. Shi, J. Mou, Q.Y. Zhuang, X.S. Wang and S.J. Tu, *Chin. J. Org. Chem.*, 2004, **24**, 1042.
- 34 D.Q. Shi, J. Mou, Q.Y. Zhuang and X.S. Wang, *Chin. J. Org. Chem.*, 2004, **24**, 1569.
- 35 D.Q. Shi, L.H. Liu, X.S. Wang, Zhang, Q.Y. Zhuang and Y. Zhang, *Acta Cryst.*, 2004, **E60**, o2371.
- 36 D.Q. Shi, L.H. Liu, X.S. Wang, Zhang, Q.Y. Zhuang and Y. Zhang, *Acta Cryst.*, 2004, **E60**, o2362.
- 37 M.Z. Piao and K. Imafuku, *Tetrahedron Lett.*, 1997, **38**, 5301.